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## Structure Reports

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## Key indicators

Single-crystal X-ray study
$T=298 \mathrm{~K}$
Mean $\sigma(\mathrm{C}-\mathrm{C})=0.006 \AA$
$R$ factor $=0.043$
$w R$ factor $=0.106$
Data-to-parameter ratio $=10.4$
For details of how these key indicators were automatically derived from the article, see http://journals.iucr.org/e.

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# 3-[3-(4-Methoxybenzoyl)thioureido]propionic acid 

In the title molecule, $\mathrm{C}_{12} \mathrm{H}_{14} \mathrm{~N}_{2} \mathrm{O}_{4} \mathrm{~S}$, the methoxybenzoyl group is trans to the thiono S atom across the $\mathrm{C}-\mathrm{N}$ bond. The propionic acid group adopts a gauche conformation about the $\mathrm{C}-\mathrm{C}$ bond, with a $\mathrm{C}_{\mathrm{COOH}}-\mathrm{C}-\mathrm{C}-\mathrm{N}$ torsion angle of 64.9 (4) ${ }^{\circ}$. In the crystal structure, intermolecular $\mathrm{O}-\mathrm{H} \cdots \mathrm{S}$, $\mathrm{N}-\mathrm{H} \cdots \mathrm{O}$ and $\mathrm{C}-\mathrm{H} \cdots \mathrm{O}$ hydrogen bonds link the molecules into chains parallel to the $c$ axis.

## Comment

The title compound, (I), is a carbonoylthiourea derivative of $\beta$-alanine analogous to 3-(3-benzoylthiouredo)propionic acid, (II) (Yusof \& Yamin, 2003), except that a methoxy group is attached at the para position of the benzoyl group (Fig. 1).

(I)

The molecule maintains its cis-trans configuration with respect to the positions of the propionic acid and 4methoxybenzoyl groups relative to the S 1 atom across the $\mathrm{C} 9-\mathrm{N} 2$ and $\mathrm{C} 9-\mathrm{N} 1$ bonds, respectively. In addition, the molecule adopts a gauche conformation about the $\mathrm{C} 10-\mathrm{C} 11$ bond, with a $\mathrm{C} 12-\mathrm{C} 11-\mathrm{C} 10-\mathrm{N} 2$ torsion angle of 64.9 (4) $\AA$; in contrast, compound (II) adopts an anti conformation with a torsion angle of $-179.12(16)^{\circ}$. The bond lengths and angles are in normal ranges (Allen et al., 1987) and comparable to those in (II). However, the O3-C12 bond length [1.325 (4) A ] is slightly longer than that in (II) $[1.296(2) \mathrm{A}]$. The 4methoxyphenyl [C1-C6/O1/C7 (A)] and central carbonylthiourea $[\mathrm{S} 1 / \mathrm{O} 2 / \mathrm{N} 1 / \mathrm{N} 2 / \mathrm{C} 8 / \mathrm{C} 9(B)]$ fragments are essentially planar. In the propionic acid fragment [O3/O4/C10/C11/C12 $(C)$ ], the maximum deviation from the mean plane is 0.186 (4) A for atom C10. The dihedral angles $A / B$ and $B / C$ are 10.35 (13) and $62.08(17)^{\circ}$, respectively. The 4-methoxyphenyl unit is inclined to the propionic acid mean plane with a dihedral angle of $54.29(19)^{\circ}$, compared with $22.46(13)^{\circ}$ in (II). The intramolecular hydrogen bond $\mathrm{N} 2-\mathrm{H} 2 A \cdots \mathrm{O} 2$

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Figure 1
The molecular structure of (I), shown with $50 \%$ probability displacement ellipsoids. The dashed line represents a hydrogen bond.


Figure 2
Packing diagram of the title complex, viewed down the $b$ axis. The dashed lines denote intermolecular $\mathrm{N}-\mathrm{H} \cdots \mathrm{O}, \mathrm{O}-\mathrm{H} \cdots \mathrm{S}$ and $\mathrm{C}-\mathrm{H} \cdots \mathrm{O}$ hydrogen bonds.
(Table 2) forms a pseudo-six-membered ring ( $\mathrm{N} 2 / \mathrm{H} 2 \mathrm{~A} / \mathrm{O} 2 / \mathrm{C} 8 /$ N1/C9).

In the crystal structure, molecules are linked by intermolecular $\mathrm{O}-\mathrm{H} \cdots \mathrm{S}, \mathrm{N}-\mathrm{H} \cdots \mathrm{O}$ and $\mathrm{C}-\mathrm{H} \cdots \mathrm{O}$ hydrogen bonds (Table 2), forming a one-dimensional chain parallel to the $c$ axis, in contrast to a two-dimensional hydrogen-bonded network in (II). The crystal packing (Fig. 2) is further stabilized by van der Waals forces.

## Experimental

A solution of 3-aminopropionic acid ( $0.05 \mathrm{~mol}, 4.45 \mathrm{~g}$ ) in 20 ml acetone was added dropwise to a two-necked round-bottomed flask containing an equimolar solution of 4-methoxybenzoylisothiocyanate $(0.05 \mathrm{~mol}, 9.65 \mathrm{~g})$ in 20 ml acetone. The mixture was refluxed for about 5 h and then filtered into a beaker containing ice cubes. The resulting yellow precipitate was washed with cold acetone and distilled water. The product was dried under vaccum and kept in a desiccator (yield $11.99 \mathrm{~g}, 85 \%$; m.p. 430.3-431.1 K). Recrystallization from ethanol yielded yellow crystals suitable for X-ray analysis.

## Crystal data

$\mathrm{C}_{12} \mathrm{H}_{14} \mathrm{~N}_{2} \mathrm{O}_{4} \mathrm{~S}$
$M_{r}=282.31$
Monoclinic, $P C$
$a=9.795$ (3) $\AA$
$b=4.7790(13) \AA$
$c=14.951$ (4) $\AA$
$\beta=108.275(5)^{\circ}$
$V=664.6(3) \AA^{3}$

$$
Z=2
$$

$D_{x}=1.411 \mathrm{Mg} \mathrm{m}^{-3}$
Mo $K \alpha$ radiation
$\mu=0.26 \mathrm{~mm}^{-1}$
$T=298$ (2) K
Plate, yellow
$0.34 \times 0.17 \times 0.07 \mathrm{~mm}$

## Data collection

Bruker SMART APEX CCD areadetector diffractometer
$\omega$ scans
Absorption correction: multi-scan
(SADABS; Bruker, 2000)
$T_{\text {min }}=0.912, T_{\text {max }}=0.979$
3159 measured reflections 1792 independent reflections 1660 reflections with $I>2 \sigma(I)$
$R_{\text {int }}=0.019$
$\theta_{\text {max }}=25.0^{\circ}$

## Refinement

Refinement on $F^{2}$
$R\left[F^{2}>2 \sigma\left(F^{2}\right)\right]=0.043$
$w R\left(F^{2}\right)=0.106$
$S=1.15$
1792 reflections
172 parameters
H-atom parameters constrained

$$
\begin{aligned}
& w=1 /\left[\sigma^{2}\left(F_{\mathrm{o}}^{2}\right)+(0.0526 P)^{2}\right. \\
& +0.0617 P] \\
& \text { where } P=\left(F_{\mathrm{o}}{ }^{2}+2 F_{\mathrm{c}}{ }^{2}\right) / 3 \\
& (\Delta / \sigma)_{\max }<0.001 \\
& \Delta \rho_{\max }=0.28 \mathrm{e}_{\AA^{-3}} \\
& \Delta \rho_{\min }=-0.15 \mathrm{e}^{-3} \\
& \text { Absolute structure: Flack (1983), } \\
& 628 \text { Friedel pairs } \\
& \text { Flack parameter: } 0.16 \text { (11) }
\end{aligned}
$$

Table 1
Selected geometric parameters ( $\left({ }^{\circ},{ }^{\circ}\right)$.

| S1-C9 | $1.687(4)$ | $\mathrm{N} 1-\mathrm{C} 9$ | $1.375(4)$ |
| :--- | :--- | :--- | :--- |
| O2-C8 | $1.231(4)$ | $\mathrm{N} 1-\mathrm{C} 8$ | $1.382(5)$ |
| O3-C12 | $1.325(4)$ | $\mathrm{N} 2-\mathrm{C} 9$ | $1.318(4)$ |
| O4-C12 | $1.201(5)$ | $\mathrm{N} 2-\mathrm{C} 10$ | $1.463(5)$ |
|  |  |  |  |
| C9-N1-C8 | $128.2(3)$ | $\mathrm{N} 2-\mathrm{C} 9-\mathrm{S} 1$ | $122.0(3)$ |
| N2-C9-N1 | $117.7(3)$ | $\mathrm{N} 1-\mathrm{C} 9-\mathrm{S} 1$ | $120.3(3)$ |

Table 2
Hydrogen-bond geometry ( $\AA{ }^{\circ},{ }^{\circ}$ ).

| $D-\mathrm{H} \cdots A$ | $D-\mathrm{H}$ | $\mathrm{H} \cdots A$ | $D \cdots A$ | $D-\mathrm{H} \cdots A$ |
| :--- | :--- | :--- | :--- | :--- |
| $\mathrm{~N} 2-\mathrm{H} 2 A \cdots \mathrm{O} 2$ | 0.86 | 1.94 | $2.622(4)$ | 135 |
| $\mathrm{O} 3-\mathrm{H} 3 \cdots \mathrm{~S} 1^{\mathrm{i}}$ | 0.82 | 2.32 | $3.126(3)$ | 169 |
| $\mathrm{~N} 1-\mathrm{H} 1 A \cdots 4^{\mathrm{ii}}$ | 0.86 | 2.26 | $3.113(4)$ | 170 |
| $\mathrm{C} 1-\mathrm{H} 1 B \cdots 4^{\mathrm{ii}}$ | 0.93 | 2.28 | $3.155(6)$ | 156 |

Symmetry codes: (i) $x,-y, z-\frac{1}{2}$; (ii) $x,-y, z+\frac{1}{2}$.

All H atoms were positioned geometrically and allowed to ride on their parent atoms, with $\mathrm{C}-\mathrm{H}=0.93-0.97 \AA, \mathrm{~N}-\mathrm{H}=0.86 \AA$ and $\mathrm{O}-$ $\mathrm{H}=0.82 \AA$, with $U_{\text {iso }}(\mathrm{H})=1.2 U_{\text {eq }}(\mathrm{C}, \mathrm{N})$ for CH and NH groups and $1.5 U_{\text {eq }}(\mathrm{C}, \mathrm{O})$ for $\mathrm{CH}_{3}$ and OH groups.

Data collection: SMART (Bruker, 2000); cell refinement: SAINT (Bruker, 2000); data reduction: SAINT; program(s) used to solve structure: SHELXS97 (Sheldrick, 1997a); program(s) used to refine structure: SHELXL97 (Sheldrick, 1997a); molecular graphics: SHELXTL (Sheldrick, 1997b); software used to prepare material for publication: SHELXTL, PARST (Nardelli, 1995) and PLATON (Spek, 2003).

## organic papers

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